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Note : Each Quantum Position As Described By The First Three Quantum Numbers Can Be Occupied By Only Two Electrons With Opposite Spins. For Example, If The Position Of A Pair Of Electrons Are Described As 4p_x, This Means They Are In The Fourth Energy Level (n=4), And Have A Dumbbell Shaped Orbital (orbital Quantum Number Is P) Around The X Axis (third Quantum Number). These Two Electrons In ... Aug 21th, 2020

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